

# Progress on new methods, materials, and insights for safety Li-ion battery systems

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### Contents

- 1. Testing and insights for safer battery systems
  - I. Understand what causes the spectrum of risks
  - II. Design testing conditions to intentionally induce the 'high-risk' failures
  - III. Using insights to improve safety of battery systems
- 2. Acoustic diagnostics for detecting failure
- 3. Materials for safer Li-ion batteries





High-risk failure mechanisms



Bursting: Top



Bursting: Bottom

#### 18650 cells



Breach: Bot

Breach: Bot



#### Hazardous flare stemming from breach



Image courtesy of E. Darcy (NASA)

Most challenging failure mechanism to handle

3

### Characterizing breaching mechanism





Breach: Top



**Cell type:** Li-ion 18650 **Capacity:** 3.4 Ah **State of Charge:** 100 % (4.2 V) **Bottom vent:** Yes **Wall thickness:** 220 μm **Orientation of cell:** Positive end up **Location of ISCD radially:** 6 winds in **Location of ISCD longitudinally:** Top **Side of ISCD in image:** Right

Location of FOV longitudinally: Top Frame rate: 2000 Hz Frame dimension (Hor x Ver): 1822 x 1140 pixels Pixel size: 10 µm

#### Cause of breach

- Reacting material fluidizes and flows towards the top vent
- Material deflects off the spingroove, causing thermal stress
- The spin-groove melts leading to a breach and escape of hot material

## Selective Positioning of the ISC Device

#### Internal short-circuiting device

• 18650 cells were manufactured with the ISC device placed at 3 different longitudinal locations



## Risk map

From a study of 200 cells, the propensity of cell to undergo certain failure mechanisms, under certain conditions, was mapped.



The number in each box represents the fraction of cells of that particular design, to undergo a particular failure mechanism

~	<b>ISC Position</b>	Design	Total	Burs	sting	Breach		Contained	
Discoloration				Тор	Bot	Тор	Side	Bot	
	None	220 µm, BV	45	0.00	1.00	0.00	0.00	0.44	0.00
	None	220 µm, NBV	46	0.02	0.00	0.07	0.02	0.16	0.98
	None	250 μm, NBV	43	0.00	0.00	0.07	0.00	0.00	1.00
6 layers in									
	Тор	220 µm, BV	11	0.09	0.27	0.64	0.00	0.27	0.64
	Mid	220 µm, BV	13	0.08	0.54	0.08	0.00	0.62	0.38
	Bot	220 µm, BV	12	0.00	1.00	0.00	0.17	1.00	0.00
6 layers in									
	Тор	250 μm, NBV	9	0.00	0.00	0.56	0.00	0.00	1.00
	Mid	250 μm, NBV	7	0.00	0.00	0.43	0.14	0.43	1.00
	Bot	250 μm, NBV	8	0.00	0.00	0.38	0.13	0.25	1.00
High risk of 3 layers in High risk of unfavorable event Finegan et al., Modelling and experiments to identify high-risk failure scenarios for testing the safety of lithium-ion cells, J. of Power Sources, <b>2019</b>									

Key findings:

- **S:** 1. Proximity of the ISC device to either end increases the risk of breach/ bursting at that end.
  - 2. Thicker casings reduce the risk of bursting but have a similar risk of breaching.
  - 3. Bottom vents reduce the risk of breaching overall, but increase the risk of bottom breaching.

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### Fractional thermal runaway calorimeter (FTRC)



#### Ejected and non-ejected heat output

- 3.6 Ah 18650 cells
- Location of thermal runaway initiation does not have significant impact on total heat output, but does influence the fraction of heat ejected
- Around **70% of heat is ejected**, mostly through the positive vent
- Initiation near the bottom increases risk of bottom breach and heat from the bottom



Finegan et al., Modelling and experiments to identify high-risk failure scenarios for testing the safety of lithium-ion cells, *J. of Power Sources*, **2019** 

### Thermal stress and bursting pressure

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#### Surface temperature and burst pressure

- The highest risk scenarios for pressure-induced breaches are when initiation of thermal runaway occurs near either end of the 18650 cell.
- Bust pressures can reach < 1.5 MPa for temperatures > 650 °C.
- If a cell produces 6 L of gas, and is clogged, the internal pressure could reach 30 Mpa..





Based on tensile strength properties for S350GD mild steel



Explains increased risk of breaching occurring, but not the consistent location at spin groove

### Results guiding safe battery designs



- Single cell data applied to battery pack simulations
  - Modelling sizing of heat sinks to avoid propagation
  - Estimating temperatures of pack enclosures when subject to ejected heat
  - Spatially quantifying the distribution of heat within an enclosure following cell failure



Work by Chuanbo Yang (NREL)

### Results guiding safe battery designs



#### NASA X57 electric aircraft



#### Eric Darcy and team at Johnson Space Center







### Battery failure databank

## **MATIONAL RENEWABLE ENERGY LABORATORY**

#### Radiography and thermal data from over 300 tests of commercial cells

- Providing engineers and researchers with data to inform models
- Link internal phenomena with external risks
- Compare heat output and mass ejection from different abuse mechanicals
  - Nail penetration
  - o Thermal abuse
  - o Internal short circuiting
- Compare different models of cells
  - Power cells
  - o Energy cells



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### **Acoustic Time of Flight - Introduction**

- Rapid in-line, non-destructive cell diagnostic technique
- High-frequency pulse-echo mechanism to evaluate electrode level changes

Attenuation  $\Delta Z = \sqrt{|\Delta \rho|K}$ Velocity

- $v = \sqrt{\frac{K}{\rho}}$
- Time of flight dependent on:
  - Material density
  - Material elastic modulus
  - Thickness
  - Structural changes



### **Acoustic Time of Flight - Introduction**



Figure 3: A simple schematic of the experimental set-up used in the Nikon XT H225 to produce simultaneous acoustic spectroscopy and X-ray imaging.



Figure 4: CAD rendering of the ultrasonic transducer in the pouch cell calorimeter cell chamber stack under compression

### **XCT – Defect Detection**



Figure 7: (a,b) Pre- and (c, d, e,) post- mortem XCT of defect driven delaminations in a commercial Li-ion cell

### **Acoustic Time of Flight – Defect Detection**





*Figure 5: Acoustic time of flight during defect induced delamination during operation of a 210 mAh pouch cell.* 

### **Acoustic Time of Flight – Thermal Runaway**





Figure 8: Selected radiography frames from Cell 1 highlighting significant structural changes during thermal runaway in the first failure test. Delamination and gas generation predicated the initiation of widespread thermal runaway.

### **Acoustic Time of Flight – Thermal Runaway**

Acoustic Amplitude



Figure 9: Acoustic time of flight during thermal runaway of Cell 1



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### **PCCs: Thermal Runaway Mechanisms**

- Anode and cathode contact via the nail causes an unmitigated short-circuit
- Elevated temperatures due to Joule heating causes thermal decomposition of electrode components, initially with the polyolefin separator and electrolyte
- In cells with the PCC, the PCC would react to the elevated temperatures before the separator could fail, this would prevent cathode-anode contact as well as electrically isolating the cathode from the nail and the anode.
- Whether the PCC's safety mechanism acted before the failure of the 10 µm polypropylene separator would dictate the efficacy of the PCC.



### **PCCs: Gravimetric Energy Density**

- Similar thicknesses to commercial metal foil current collectors (CCs):
  - Al CC: 15 μm, Al PCC: 11 μm
  - Cu CC: 10 μm, Cu PCC: 11 μm
- PCCs have a polymer substrate (ca. 8  $\mu m$  thickness) with ca. 0.5  $\mu m$  metal film coating of Al or Cu
- Therefore, a significant reduction in the amount of metal required by the PCC compared to commercial metal foils used as current collectors
- This reduction in metal is noticeable on the cell level as the average mass reduction was 2.2 grams, ca. 5% of total mass of a commercial metal foil CC control cell
- Mechanical properties of PCCs allow for assimilation with current cylindrical cell roll-to-roll manufacturing technology



### **PCCs: Cell Groups and CT**





### **Results: Failure Prevention and Calorimetry**

- Mass reduction in cells with one or both of the PCCs, mass reduction of 2.2 g with both
- Significant difference in observed TR output energy, maximum nail TC temperature and net cell mass loss with Al PCC cells
- Cu PCC alone did not protect the cells from undergoing thermal runaway but mitigation of the output energy was observed. This was due to the nail in contact with the cell during penetration, thus the Cu PCC despite protecting the anode, the graphite was connected to the cell can and shorted with the unprotected cathode.



### **Results: Synchrotron X-ray Radiography**

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#### (a) G4-01 (Al CC + Cu CC) Radiography



#### Characteristic cracking when Al PCC is absent

#### **(b)** G1-01 (Al PCC + Cu CC) Radiography



Electrode layers adjacent to nail splitting

### **Results: Post-mortem CT and 4.077 V**



Sub-micron X-ray CT reveal AI PCC had shrunk away from the elevated temperature of the nail, which caused the microscopic short-circuit. This prevented further short-circuiting, subsequent OCV measurement showed 4.077 V. Cells retained voltage for over 10 months.

### Conclusions



- High-speed X-ray imaging useful for guiding and validating thermal runaway models for identifying internal and external hotspots.
- Highest surface temperatures and lowest burst pressures were achieved when initiation occurred near either ends of the cell, due to relatively poor heat dissipation.
- The likelihood of high-risk failure scenarios can be increased by selectively locating the point of thermal runaway initiation within a cell.
- Thermal data from the fractional thermal runaway calorimeter (FTRC) is useful for accurately modelling efficacy of heat sinks and enclosures for withstanding thermal runaway.
- An **open-source database** of radiography and thermal data to be released soon.
- Acoustic technique demonstrated to enable detection of deformed electrode layers leading up to thermal runaway.
- Soteria's polymer substrate current collectors demonstrated to withstand nail penetration without thermal runaway where otherwise thermal runaway consistently occurs. Light-weight, cost effective, and mechanically robust for scaling up.

### Thank you for listening

#### **Donal Finegan**

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#### List of relevant publications:

- **1.** Finegan et al., Characterising thermal runaway within lithium-ion cells by inducing and monitoring internal short circuits. *Energy & Environmental Science* **2017**, *10* (6), 1377-1388.
- **2.** Finegan et al., Identifying the Cause of Rupture of Li-Ion Batteries during Thermal Runaway. *Advanced Science*, 1700369, **2017**.
- **3.** Finegan et al., In-operando high-speed tomography of lithium-ion batteries during thermal runaway. *Nature Communications* **2015**, *6*.
- **4.** Finegan et al., Tracking Internal Temperature and Structural Dynamics during Nail Penetration of Lithium-Ion Cells. *Journal of The Electrochemical Society* **2017**, *164* (13), A3285-A3291.
- **5.** Finegan et al., Modelling and experiments to identify high-risk failure scenarios for testing the safety of lithium-ion cells, *J. of Power Sources*, **2019**
- **6.** Walker et al., Decoupling of heat generated from ejected and non-ejected contents of 18650-format lithium-ion cells using statistical methods, *J. Power Sources*, **2019**
- **7.** Robinson et al., Spatially Resolved Ultrasound Acoustic Investigations of Li-Ion Batteries, *Phys. Chem. Chem. Phys.*, **2019**
- **8.** Pham et al., Correlative acoustic time-of-flight spectroscopy and X-ray imaging to investigate gasinduced delamination in lithium-ion pouch cells during thermal runaway, *J. Power Sources*, **2020**

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