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Scalable, High Energy Density Lithium-Sulfur Batteries (SD-LSB)

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NASA's energy storage needs for future space missions

 NASA JPL Whitepaper "Energy Storage Technologies for Planetary Science and Astrobiology Missions". May 01, 2021*

Table 3: Performance Characteristics of Emerging Rechargeable Battery Technologies							
Technology	Projected Performance (75% of Cell level)				Challenges	Kev Players	TRL
	Wh/kg	Wh/l	Cycle Life	Temp	chancinger		
Li-S with Liquid Ely'te	325	600	200	-20 to +50	Sulfur shuttle, Li dendrites, Safety?	Oxis Energy, Sion Power ,	3

325 Wh/kg (battery level) \rightarrow 433 Wh/kg (cell level)

* Bugga, R.; Brandon, E.; Darcy, E.; Ewell, R.; Faguay, P.; et al. "Energy Storage Technologies for Planetary Science and Astrobiology Missions". May 01, 2021.



Li-S battery advantages

- High theoretical specific capacity / energy S half redox reaction:
 - $S_8 + 16Li^+ + 16e^- \leftrightarrow 8Li_2S$ or
 - $S + 2Li^+ + 2e^- \leftrightarrow Li_2S$ (Li₂S = lithium sulfide)
 - S theoretical specific capacity 1,675 mAh/g
 - Li-S battery theoretical specific energy ~ 2,500 Wh/kg
- Naturally abundant
- Low cost
- Environmentally friendly



Li-S battery challenges



 S cathode and Li form a series of intermediates (Li₂S_n n>1) before the final product (Li₂S)



Li-S battery challenges

- Dissolution of high-order Li polysulfides (LPS), Li_2S_n (4 \le n \le 8)
 - Diffusion of LPS anions (S_n²⁻) through the separator to the negative Li anode can cause
- LPS shuttle phenomenon during charge
 - High order LPS diffuses to the negative electrode and reacts with Li anode to form low order LPS
 - Low order LPS diffuses back to the positive electrode to be oxidized to form high order LPS
- Poor electronic conductivity of S and low order LPS
 - \circ S is 5 x 10⁻³⁰ S/cm at 25°C; (compare Cu 6 x 10⁷ S/cm)
 - \odot Low order LPS (Li₂S₂, Li₂S) are insulating in nature
- Sulfur electrode volume change
 - \circ ~76% volume change during cycling \rightarrow loss of particle contact
- Use of metallic Li anode
 - \circ Dendrite formation \rightarrow safety concerns

M SOUTH DAKOTA MINES Nanolayer polymer modified carbon (NPC)



- NPC
- Renewable biomass corn stover starting material
- Final product: **NPC**.



Li-S battery preparation

- NPC modified sulfur electrodes: NPC-S.
- Baseline sulfur electrodes: Control
- Li anodes
- Electrolyte: LiTFSI in DOL/DME
- Polyolefin separators
- CR2032 coin-type cell hardware

Specific capacity of Li-S cells



- NPC-S cell delivered discharge specific capacity ~1,600 mAh/g which approaches the theoretical value of 1,672 mAh/g
- The control cell delivered only 800 mAh/g due to adverse effect of PS formation resulting in loss of active sulfur

Cycle life of Li-S cells

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• The NPC-S based cell delivered twice the specific capacity than the baseline-S cell (940 mAh/g vs. 470 mAh/g) after 100 cycles

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In-situ Raman of Li-S cells



- The baseline-S showed characteristic PS peaks due to dissolution of PS species into the electrolyte
- The NPC-S showed no PS peaks (no PS dissolution into the electrolyte)



- The Carbon-S and Polymer-S cells delivered significantly lower discharge specific capacities, 580 mAh/g and 140 mAh/g, respectively, than 1,600 mAh/g of the NPC-S cell.
- No or inadequate PS trapping capability and large charge transfer interfacial impedance led to the low discharge specific capacities of the Carbon-S and Polymer-S cells

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Molecular dynamics simulation



- Interfacial charge distribution: red: electron gain green: electron loss
- Accumulation of opposite charges on NPC leads to the formation of surface dipoles.
- MD simulation suggests NPC's PS trapping capability arises from dipole-dipole interactions



Conclusions

- A new type of PS-trapping material, NPC, was studied
- NPC-S based Li-S cells demonstrated near theoretical discharge specific capacities
- Our novel approach is simple, low-cost, and economically scalable for large-scale commercialization
- SD-LSB holds promise to address NASA battery energy storage needs for future space missions

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