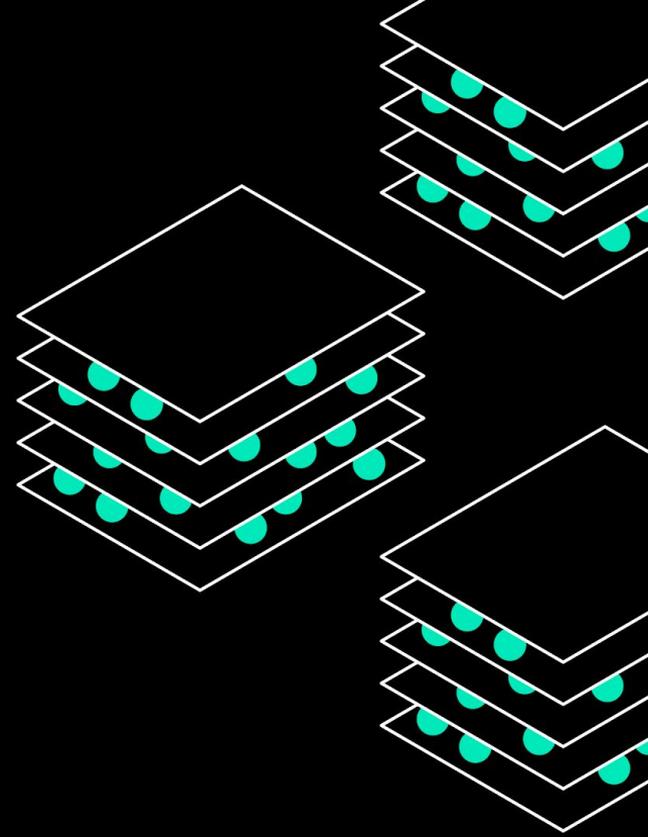


VOLTAIQ

Enhanced COTS Cell Qualification with Electrochemical Pulsing Protocols

Blake Hawley, PhD
Senior Battery Engineer
2024 NASA Aerospace Battery Workshop
November 20, 2024



Voltaiq software is **production-proven** at leading companies spanning the battery value chain:

Founded in 2012, Voltaiq pioneered the battery intelligence category, and remains the leader in battery quality analytics software today. Our robust solution and world-class team of battery experts is ready to help accelerate our customers to success.



Mercedes-Benz



FREYR



UNIVERSITY OF MICHIGAN



element

PEAK ENERGY

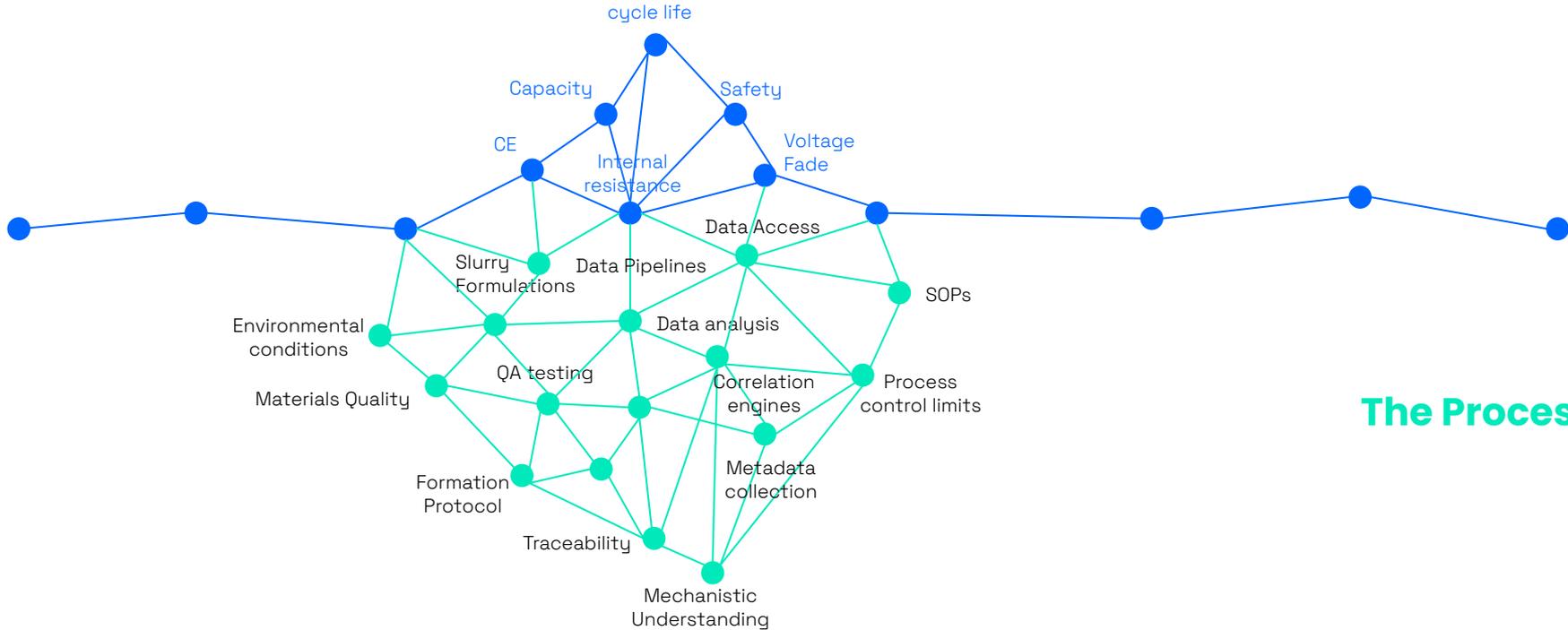


ANOVION™



The Battery Iceberg: Performance or Process

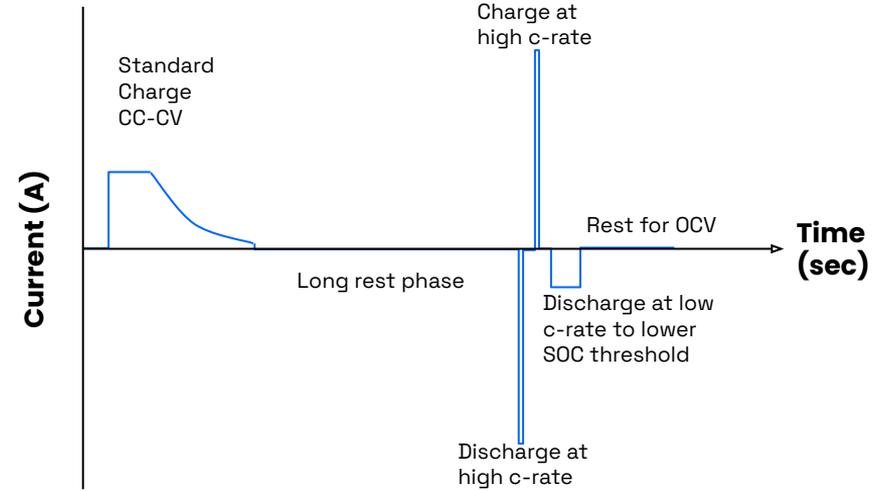
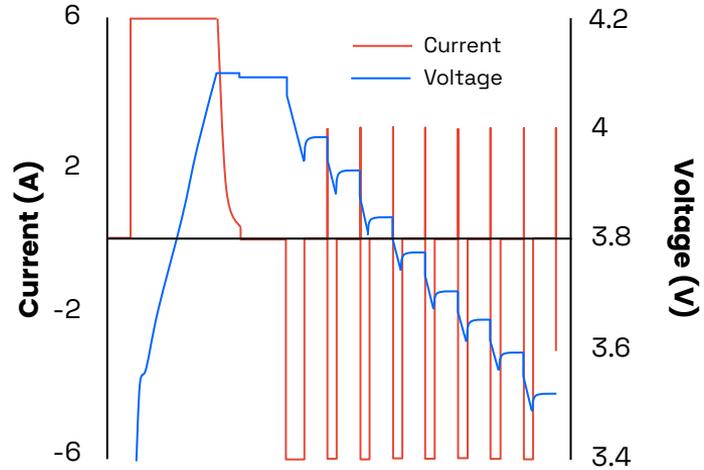
The Performance



The Process



HPPC lacks connection to fundamental phenomena



Analyzing HPPC data via Equivalent circuit

Current (A) vs Time (sec) showing a step change ΔI .

Voltage (V) vs Time (sec) showing a step change with three distinct voltage drops: Δv_0 , Δv_1 , and Δv_2 .

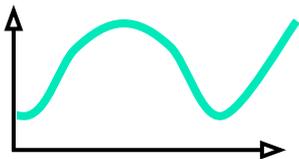
Equivalent circuit diagram showing a voltage source V_{oc} , a resistor R_0 , and two parallel RC branches (R_1, C_1 and R_2, C_2) connected to the cell terminals V_{cell} .

$R_x = \Delta v_x / \Delta I$



EIS complements DC pulsing, but is expensive and hard to scale

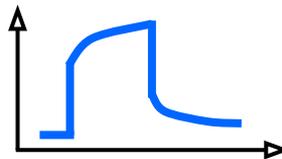
Electrochemical Impedance Spectroscopy



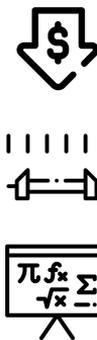
- Uses **alternating current** and frequency response analyzer
- Empirical analysis established
- Clearly **separates out time dependent phenomena**
- Testing at different voltages difficult
- Must be used **only at small currents**



Pulse Testing

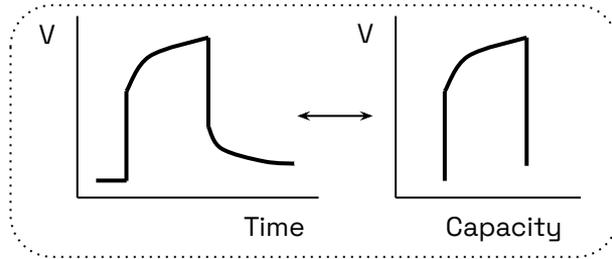


- Uses direct current and no special equipment
- Empirical analysis not obvious
- Does not clearly separate out time dependant phenomena
- Can test a range of currents
- Testing at different voltages is straightforward



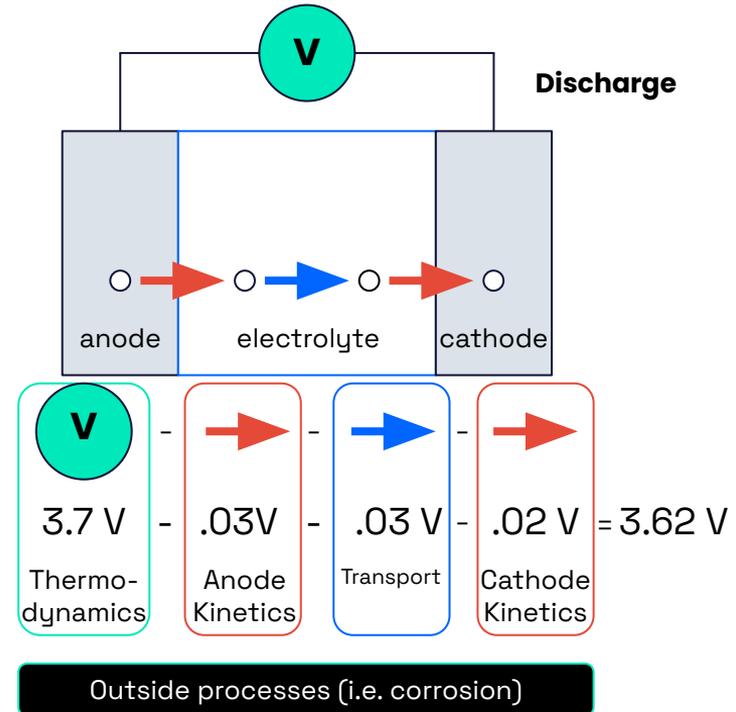
Voltage changes inform us of fundamental phenomena

Thermodynamics, kinetics, and transport

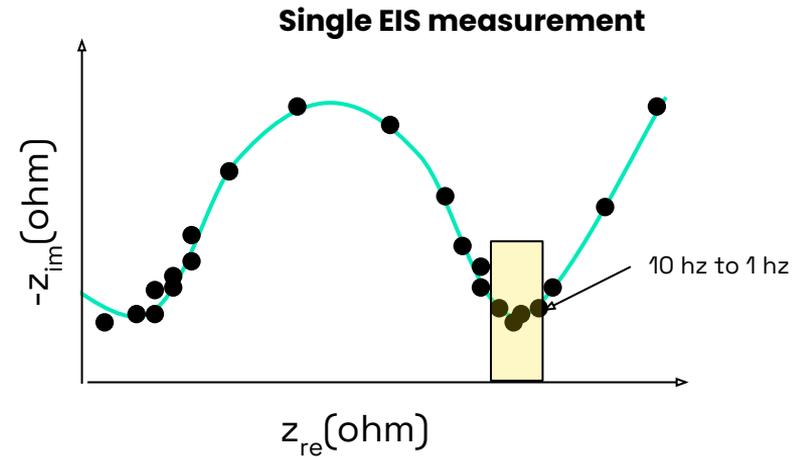
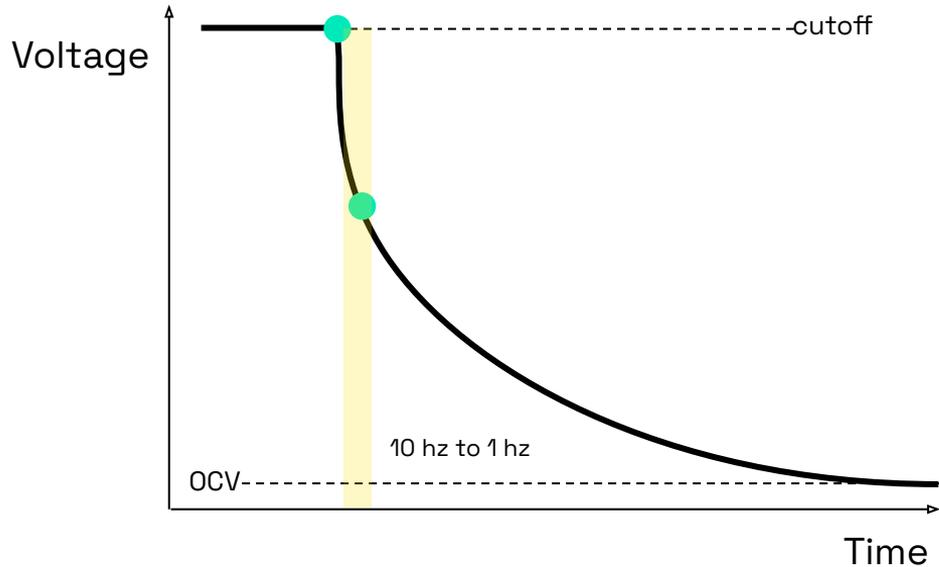


Voltage can change due to:

1. Changes in standard electrode potential
2. Anode kinetics
3. Transport phenomena
4. Cathode kinetics
5. Outside processes

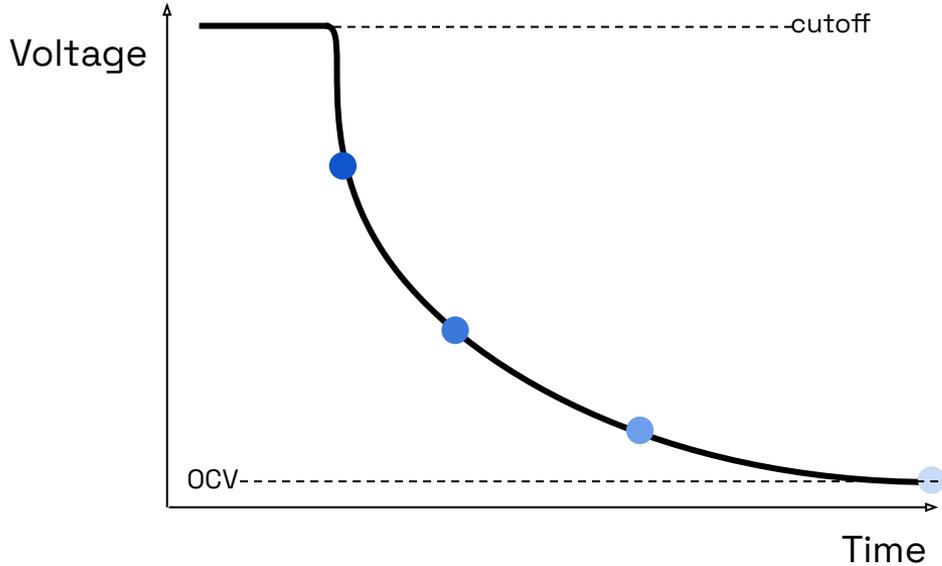


Kinetics account for initial change after current interruption

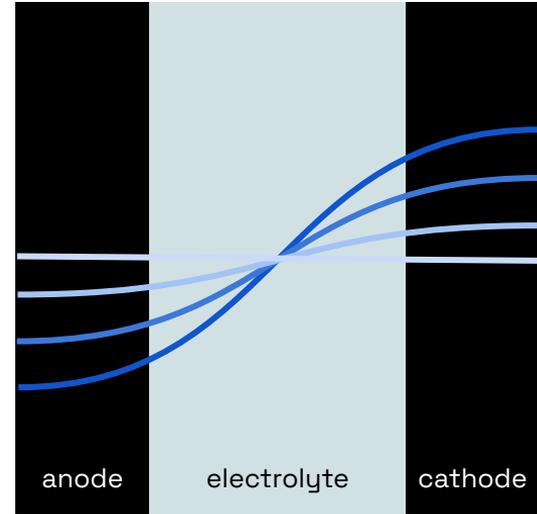


- Some potential must be sacrificed at each electrode to move an ion from electrolyte to electrode
- This potential change is removed after the current is removed

Transport accounts for change after initial drop

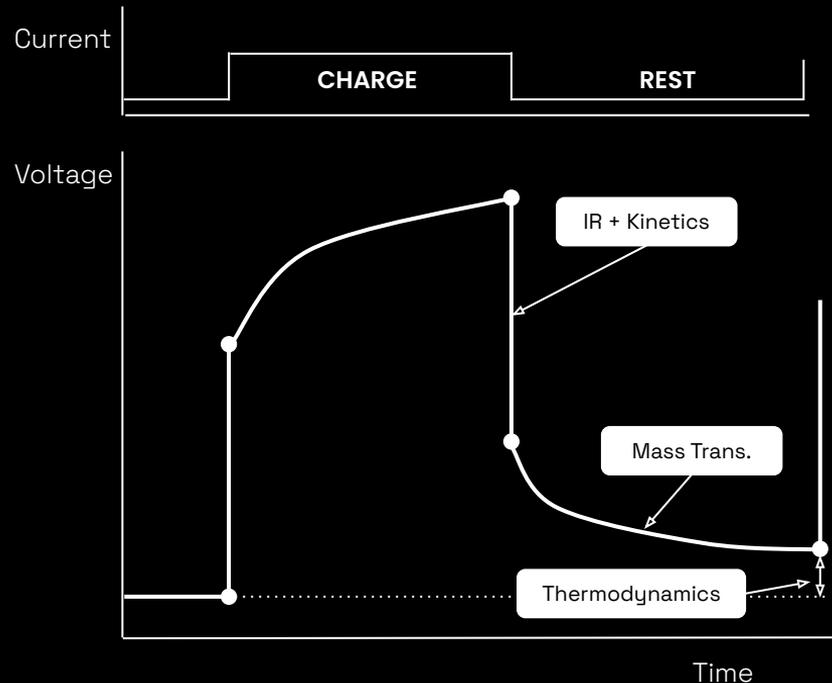
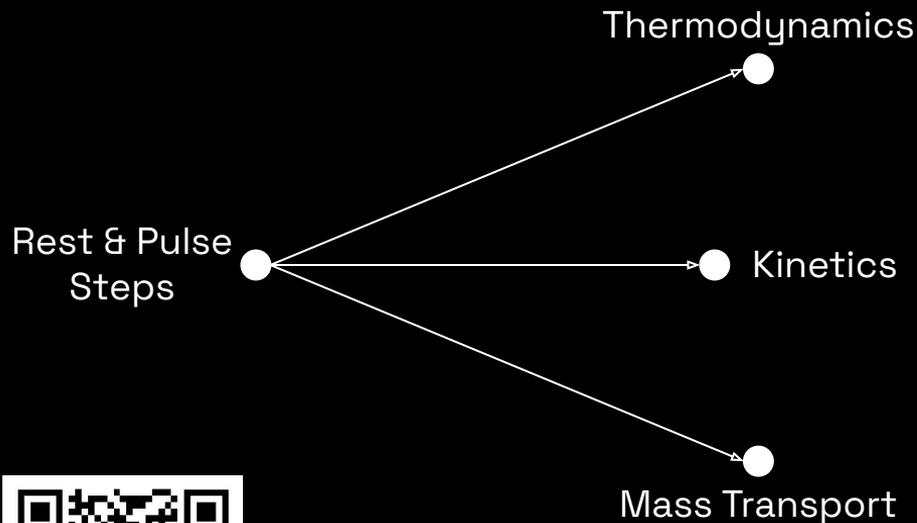


Concentration Across Cell



- After current perturbations are removed Li-ions can move back to equilibrium
- The change in concentration affects cell voltage

A better way towards mechanistic understanding



The pulse is not new but the *ANALYSIS and STRUCTURE* is...

APP enables efficient gathering of maximal information

Capacity-Controlled Pulses

Allows for constant dQ step in s -DQDV

Multiple Currents

Allows for fitting to Butler-Volmer kinetics, understanding how resistance shifts with current

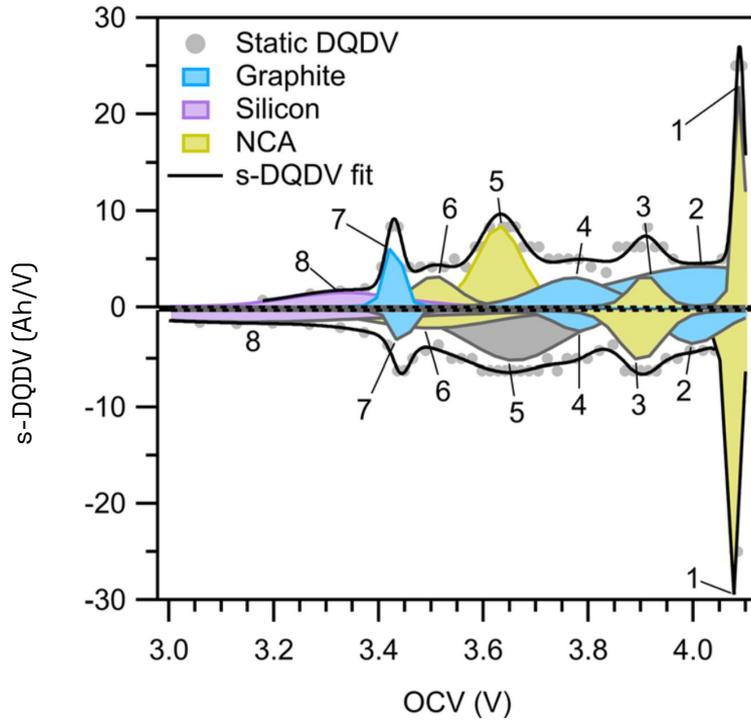
Optimized Sampling

High-frequency sampling during pulse and immediately after interruption, low-frequency elsewhere

dV/dt End Condition for Rest

Avoids unnecessarily long rest times when cell has already reached equilibrium

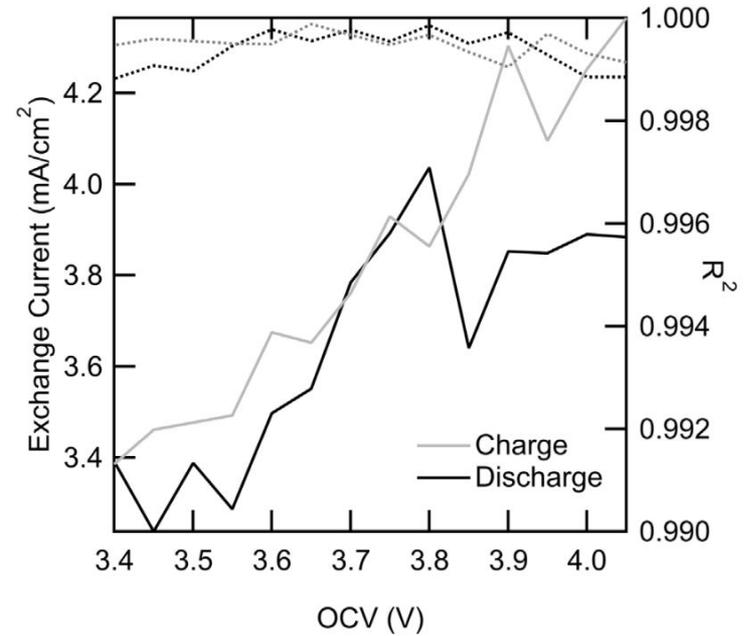
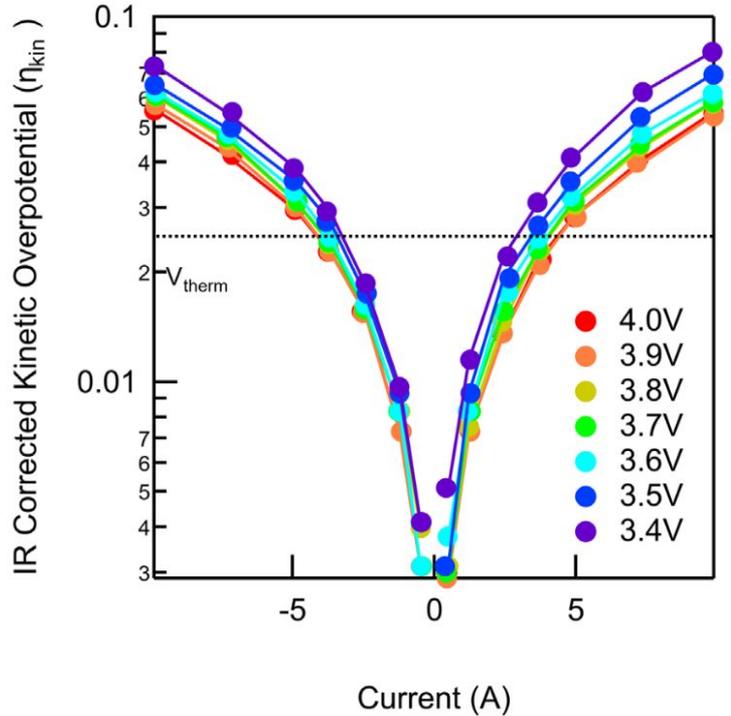
s-DQDV with Gaussian fit, dQ/dV half cell data for full mapping



Feature	Peak 1	Peak 2	Peak 4	Peak 5	Peak 8
Voltage Position Charge (V)	4.09	4.02	3.77	3.63	3.34
Voltage Position Discharge (V)	4.08	4.01	3.78	3.66	3.15
ΔG_{rxn} (kJ/mol)	-394.5	-387.9	-363.7	-350.3	-321.8
Dominant Cathode Rxn	H2 <> H3	H2	M	H1 <> M	H1
Dominant Anode Rxn	Stage II <> Stage I	Stage II <> Stage I	Stage III <> Stage II	Stage III <> Stage II	Si

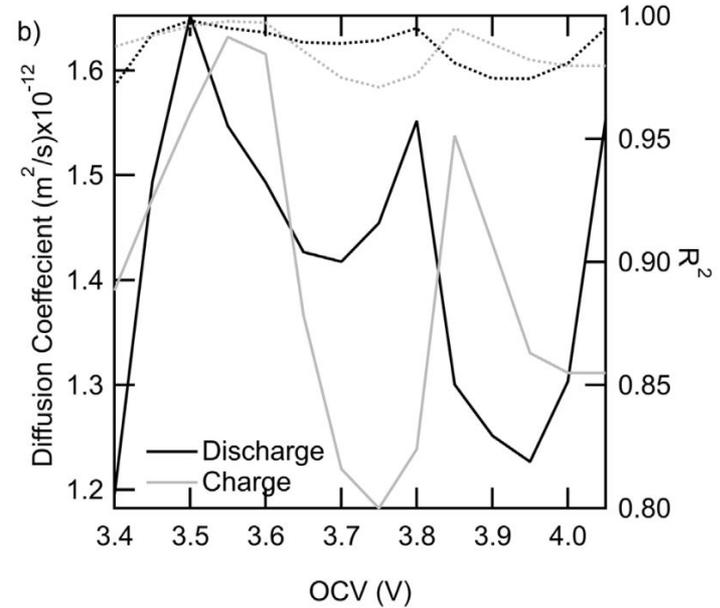
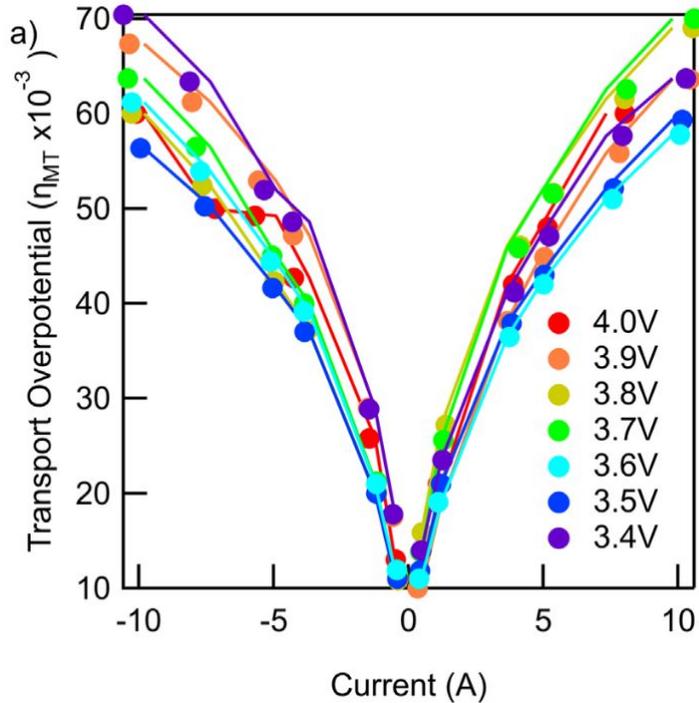
$$\Delta G_{rxn} = -nFV_{peak}$$

Pulsing unlocks non-linear Butler-Volmer kinetics



$$j = j_0 \left[A e^{\frac{(1-\alpha)nF\eta_{kin}}{RT}} - B e^{\frac{\alpha nF\eta_{kin}}{RT}} \right]$$

Crossover in diffusion coefficient correlate with phase changes

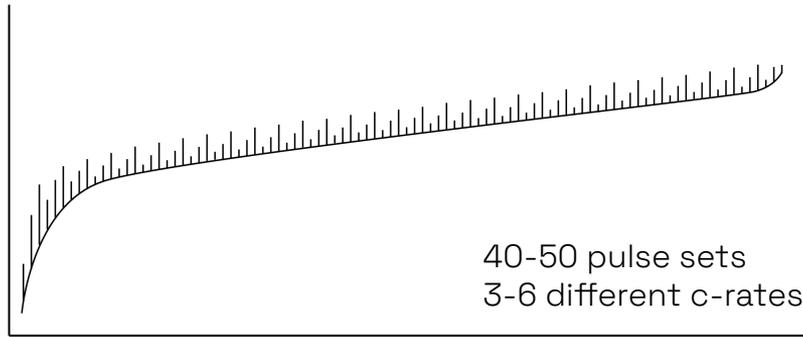


$$j = \frac{Cz^2F^2}{2RT} \sqrt{\frac{nD}{t}} \eta_{mt}$$

APPs for Cell Qualification or Rapid Benchmarking

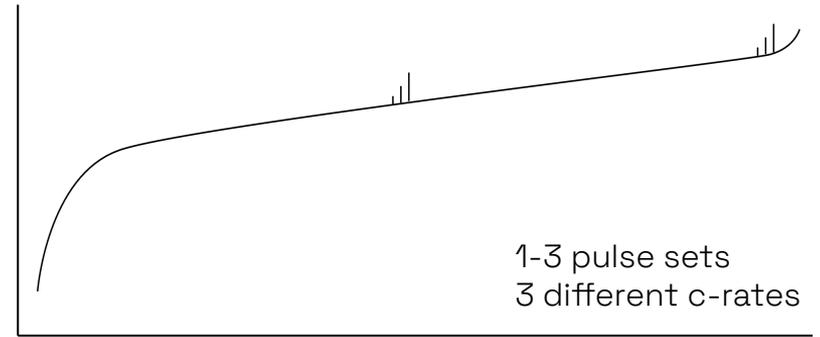
Analytical Pulsing Profiles (APPs) can be adapted to time and data constraints to provide full mechanistic insight

Cell Qualification



- Graphical Analysis
- Cell level mechanistic information
- Understanding of chemistry and process

Rapid Benchmarking



- Device level information
- Identify outliers mechanistically
- Understand process deviations

VOLTAIQ

Thank you!

Blake Hawley, PhD
Senior Battery Engineer
blake.hawley@voltaiq.com

